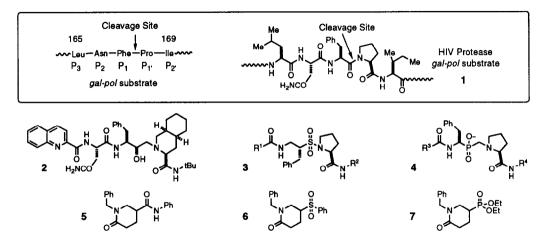
"Conformationally Restricted β-Amino Acid Isosteres Prepared Through Regioselectively Controlled Aza-Annulation."

K. Paulvannan and John R. Stille* Department of Chemistry, Michigan State University, East Lansing, MI 48824

Abstract: A variety of electron withdrawing substituents were used to enhance the aza-annulation of enamines with acryloyl chloride, to direct the regioselectivity of alkene formation, and to facilitate hydrogenation of the unsaturated annulation product. The resulting δ -lactam products were β -amino acid analogs with structural features similar to those of established peptide isosteres.

Many important biological processes rely on hydrogen bonding as the key to substrate binding and transition state stabilization. For peptide substrates, specificity of molecular recognition is dependent upon the functional groups present, and their relative topological arrangement. In rational drug design, each of these factors is critical in the evolution of small molecules into natural substrate mimics that will attenuate targeted biological processes through competitive inhibition or enzyme inactivation. For example, this chemotherapeutic strategy has been used in the inhibition of HIV protease, which is essential for viral maturation and infectivity.¹ Peptide isosteres specifically designed for inhibition of this enzyme include 2,² 3,³ and 4,⁴ and in each case, these peptide mimics contain a 1,3 substitution pattern of nitrogen or oxygen heteroatoms. In order to design greater structural control into these amino acid analogs, synthetic routes to compounds related in structure to 5, 6, and 7 have been investigated. This work represents an approach to the synthesis of conformationally restricted β -amino acid isosteres prepared through regioselectively controlled aza-annulation of enamines with acryloyl chloride.



A number of research programs have focused their efforts on preparing peptidomimetic molecules.⁵ Several approaches achieved conformational control of the backbone through the incorporation of ring systems,⁶ which controlled the spatial arrangement of functional groups by substitution pattern, stereochemistry, and rotational constraints. In these studies, nonpeptidal peptidomimetics have been prepared that imitate naturally occurring β -pleated sheet conformations⁷ and β -turns,⁸ in which cyclic systems constrain the conformations of 1,2- and 1,3-heteroatom relationships. Other peptide analogs, which have greater potency and increased enzymatic stability, have been prepared by substitution of β -amino acids for α -amino acids in biologically active peptides.⁹ Herein, a convergent aza-annulation approach to the synthesis of δ -lactams, through stepwise *C*-alkylation (conjugate addition) and *N*-acylation, is described for the preparation of 1,3-heteroatom substitution patterns similar to β -amino acid derivatives and isosteres 5, 6, and 7.

The placement of an electron withdrawing group (EWG) in a position β to the ketimine has several key advantages in the aza-annulation reaction (eq. 1). In the absence of an electron withdrawing group, such as reported for the benzyl imine of cyclohexanone, treatment with acryloyl chloride produced a mixture of the *N*-acylation product (15%) and aza-annulation products (85%) as an 85:15 mixture of two olefin regioisomers.¹⁰ In the presence of an electron withdrawing group, however, the tautomeric equilibrium shifts from the ketimine form to that of β -enamino functionality. Enhancement of the enamine form produced significant increases in both reaction yield and selectivity under similar aza-annulation conditions.¹¹

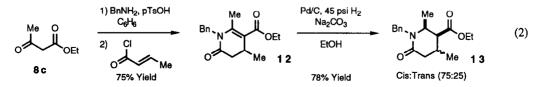
H ₂ , pTsOH	X EWG	Pd/C, 45 psi H ₂ Na ₂ CO ₃ EtOH	Bn X EWG +	Bn, N, EWG (1)
0	∽ ₉		0 10	0~ 11
Substrate (8)		Iso	lated Yield (%)	Selectivity
X	EWG	9	(10+11)	(10:11)
Me	C(O)Me	84	87	72:28
Me	C(O)Ph	96	see text	
Me	CO ₂ Et	94	91	>99:1
CO ₂ Me	CO ₂ Me	74	80	>99:1
-CH2OC(O)-		48	83	>95:5
Me	C(O)NHP	n 53	56	>95:5
Me	P(O)(OEt)	2 72	67	78:22
Me	SO ₂ Me	45	see text	
Me	SO ₂ Ph	69	see text	
	Bn N Bn N O Substrate (X Me Me CO2Me -CH2 ^t Me Me Me Me Me	Bn N Bn N Bn N Bn N Br EWG 9 Substrate (8) X EWG Me C(0)Me Me C(0)Ph Me CO2Et CO2Me CO2Me -CH2OC(0)- Me C(0)NHPI Me P(0)(OEt); Me SO2Me	Bn EWG Na2CO3 Bn EWG EtOH 9 EtOH Iso Substrate (8) Iso X EWG 9 Me C(O)Me 84 Me C(O)Ph 96 Me CO2Et 94 CO2Me CO2Me 74 -CH2OC(O)- 48 Me C(O)NHPh 53 Me P(O)(OEt)2 72 Me SO2Me 45	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

Table. Aza-Annulation of β -Enamino Derivatives and Subsequent Hydrogenation to δ -Lactams.

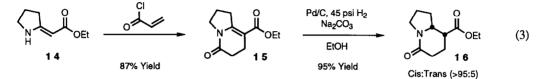
The reaction of β -diketones **8a** and **8b** in the two-step enamine condensation/aza-annulation procedure produced **9a** and **9b** in excellent yields. In each case, a single isomeric alkene product was generated, and complete regioselectivity was observed for enamine formation and annulation of the unsymmetrical β -diketone **8b** (eq. 1, Table).¹² Hydrogenation of the tetrasubstituted alkene gave a mixture of diastercomers for **9a**, but generated three products upon reduction of **9b**, including partial reduction of the benzylic carbonyl.

The use of ester EWG substituents also resulted in efficient aza-annulation, but in contrast to the reduction of **9a** and **9b**, hydrogenation of **9c** led to highly *cis* selective product formation (eq. 1, Table).^{11,12}

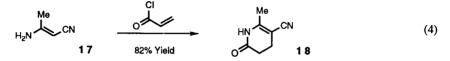
Compound 9d was formed by a slight modification of this two-step procedure, in which the enamine was formed by the addition of PhCH₂NH₂ to MeO₂CC=CCO₂Me, and then treatment with acryloyl chloride gave 9d. Diastereoselective formation of 10d was observed upon subsequent hydrogenation. The reaction of 8e under identical conditions resulted in a lower yield of 9e, but hydrogenation still selectively and efficiently produced 10e. Rationale for the different selectivities observed for 9a and 9c was the less favorable enolization of the position α to an ester relative to that of a ketone. The reaction of β -keto esters was also extended to aza-annulation with crotonyl chloride, which produced 12 in 75% yield from 8c (eq. 2). Hydrogenation of 12 exhibited 75:25 facial selectivity based on the existing stereogenic center.



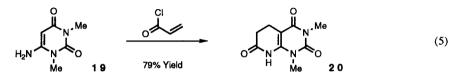
Aza-annulation of the cyclic enamino ester 14, a homolog of proline, produced 15, the ester of a β -amino acid (eq. 3).¹³ The resulting indolizidine product 15 provides some interesting possibilities for amino acid derivatives, and has been used as an intermediate in the synthesis of (±)-tashiromine.¹⁴



Turning our attention to the formation of isosteres related to the biologically active pharmaceuticals 2-4, the condensation/aza-annulation sequence was studied for substrates 8f-i (eq. 1, Table). The aza-annulation and hydrogenation of 8f resulted in slightly lower yields than those observed for the esters, but *cis* selectivity in the reduction was high (>95:5). The use of the phosphorous and sulfur¹⁵ electron withdrawing heteroatom substituents gave mixed results. Compound 9g was efficiently prepared by the two-step convergent process, and hydrogenation led to a 78:22 ratio of diastereomers. For compounds 8h and 8i, the sulfone substituent showed a significant effect on the aza-annulation process, with the Ph substituent producing more favorable results. Hydrogenation of 9h produced an undefined mixture of products, while the reduction of 9i did not proceed under these conditions. In a related example, the reaction of 17 with acryloyl chloride produced formation of 18 through aza-annulation in good yield, despite the presence of the primary NH₂ enamine (eq. 4). Compound 18 could not be reduced under the standard hydrogenation conditions.



An interesting extension of this chemistry, which further demonstrates the utility of this methodology, is aza-annulation with the amino uracil derivative **19** (eq. 5). In this case, aza-annulation efficiently produced the nucleic acid analog **20**.



In summary, the aza-annulation of stabilized enamino substrates with acryloyl chloride provides an efficient and highly regioselective method for preparing β -amino acid derivatives and peptide isosteres. In some cases, these unsaturated lactams were stereoselectively reduced by catalytic hydrogenation, which was complementary to the *trans* selectivity observed by others for ionic reductions.

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